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Chemistry Solutions Practice Problems 1. Molar solutions. a. Describe how you would prepare 1 L of a 1 M solution of sodium chloride. The gram formula weight of sodium chloride is 58.44 g/mol. Answer: To make a 1 M solution of sodium chloride, dissolve 58.44 g sodium chloride in 500 mL water in a 1000-mL volumetric flask. When all the solid is dissolved and the solution is at room

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temperature, dilute to the mark and invert the flask several times to mix.

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PROBLEM \\(\PageIndex{1}\) How do solutions differ from compounds? From other mixtures? Answer . A solution can vary in composition, while a compound cannot vary in composition. Solutions are homogeneous at the molecular level, while other mixtures are heterogeneous.

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Worked Chemistry Problem Examples - ThoughtCo

Solutions. Study Questions; Answers. More Study Questions; Answers. Worksheet of Molarity

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Problems from the ChemTeam. Worksheet of Problems from the ChemTeam on density, mass percent, molality and molarity. These problems have the answers worked out in detail. Practice Problems on Molarity from ChemTutor. animation of salt dissolving

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Review of Dilution, Concentration, and Stock Solutions . A dilution is a solution made by adding more solvent to a more concentrated solution (stock solution), which reduces the concentration of the solute. An example of a dilute solution is tap water, which is mostly water (solvent), with a small amount of dissolved minerals and gasses (solutes).

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Dilution Calculations From Stock Solutions in Chemistry

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1. A 0.750 L aqueous solution contains 90.0 g of ethanol, C_2H_5OH . Calculate the molar concentration of the solution in $mol \cdot L^{-1}$.: Solution:

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