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### Solutions Manual

Chapter 21: Chapter Outline. The following annotated chapter outline will help you review the major topics covered in this chapter. Instructions: Review the outline to recall events and their relationships as presented in the chapter. Return to skim any sections that seem unfamiliar.

### Chapter 21: Chapter Outline - Textbooks & Resources for ...

SOLUTIONS FOR CHAPTER 21 PROBLEMS 4, 12, 19, 25, 33, 40, 50, 75, 89, 96. 21.4.IDENTIFY: Use the mass  $m$  of the ring and the atomic mass  $M$  of gold to calculate the number of gold atoms. Each atom has 79 protons and an equal number of electrons. SET UP:  $23 N_A 6.02 \cdot 10^{23}$  atoms/mol. A proton has charge  $+e$ .

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