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protons; neutrons 2. electrons 3.  
neutrons 4. electrons 5. ionic 6. The two  
main types of chemical bonds are ionic  
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18.4 Structure and General Properties of the Nonmetals; 18.5 Occurrence, Preparation, and Compounds of Hydrogen; 18.6 Occurrence, Preparation, and Properties of Carbonates; 18.7 Occurrence, Preparation, and Properties of Nitrogen

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This chapter will lay the foundation for



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our study of the language of chemistry. The concepts of this foundation include the atomic theory, the composition and mass of an atom, the variability of the composition of isotopes, ion formation, chemical bonds in ionic and covalent compounds, the types of chemical reactions, and the naming of compounds.

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Cells & Cellular Processes: Chapters  
7-10. Unit 4 Genetics: Chapters 11-15.  
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Chapters 16-19. Unit 6: Viruses,  
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Chapter 2: Introduction to the Chemistry of Life Figure 2.1 Foods such as bread, fruit, and cheese are rich sources of biological macromolecules. The elements carbon, hydrogen, nitrogen,

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oxygen, sulfur, and phosphorus are the key building blocks of the chemicals found in living things.

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This video series introduces Chemistry to Anatomy and Physiology students. It covers atoms, elements, subatomic

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particles, chemical bonds, and chemical reactions. There are 3 videos in the series ...

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Biology Chapter 2- The Chemistry of Life.  
Essential Question: What are the basic  
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