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*Page 3/27*

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Chemical Formulas  
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CHAPTER 7 REVIEW  
Chemical Formulas and  
Chemical Compounds

SECTION 2 SHORT

ANSWER Answer the  
following questions in  
the space provided. 1.

Assign the oxidation  
number to the

specified element in  
each of the following

examples: 4 a. S in H

2SO 3 6 b. S in MgSO 4

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Chemical Formulas and Compounds

2 c. S in  $K_2S$  1 d. Cu in  $Cu_2S$  6 e. Cr in  $Na_2CrO_4$  5 f. N in  $HNO_3$  3  
4 g. C in  $(HCO_3)_3$  3 h. N in  $(NH_4)_2SCl_2$  2 a.

## **7 Chemical Formulas and Chemical Compounds**

CHAPTER 7 REVIEW

Chemical Formulas and  
Chemical Compounds

MIXED REVIEW SHORT

ANSWER Answer the  
following questions in  
the space provided. 1.  
Write formulas for the

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Chemical  
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following compounds:  
CuCO<sub>3</sub> a. copper(II)  
carbonate Na<sub>2</sub>SO<sub>3</sub> b.  
sodium sulfite (NH<sub>4</sub>)  
<sub>3</sub>PO<sub>4</sub> c. ammonium  
phosphate SnS<sub>2</sub> d.  
tin(IV) sulfide HNO<sub>2</sub> e.  
nitrous acid 2.

## **7 Chemical Formulas and Chemical Compounds**

CHAPTER 7 REVIEW  
Chemical Formulas and  
Chemical Compounds  
SECTION 2 SHORT  
ANSWER Answer the

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Chemical Formulas and Compounds

following questions in the space provided. 1. Assign the oxidation number to the specified element in each of the following examples: a. S in  $\text{H}_2\text{SO}_3$  b. S in  $\text{MgSO}_4$  c. S in  $\text{K}_2\text{S}$  d. Cu in  $\text{Cu}_2\text{S}$  e. Cr in  $\text{Na}_2\text{CrO}_4$  f. N in  $\text{HNO}_3$  g. C in  $(\text{HCO}_3)$  h. N in  $(\text{NH}_4)$  2. a.

## Chemical Formulas and Chemical Compounds -

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## **MAFIADOC.COM**

Answers - Naming  
Chemical Compounds .

Name the following  
chemical compounds:

- 1) NaBr sodium bromide.
- 2)  $\text{Ca}(\text{C}_2\text{H}_3\text{O}_2)_2$  calcium acetate.
- 3)  $\text{P}_2\text{O}_5$  diphosphorus pentoxide.
- 4)  $\text{Ti}(\text{SO}_4)_2$  titanium(IV) sulfate.
- 5)  $\text{FePO}_4$  iron(III) phosphate.
- 6)  $\text{K}_3\text{N}$  potassium nitride.
- 7)  $\text{SO}_2$  sulfur dioxide.
- 8)  $\text{CuOH}$  copper(I)



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hydroxide. 9)  $\text{Zn}(\text{NO}_3)_2$   
Formulas

## **Answers - Naming Chemical Compounds**

Review- Naming  
Chemical Compounds

The following are a good mix of naming and formula writing problems to help you get some practice. Name the following chemical compounds:

## **Review- Naming Chemical**

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CHAPTER 7 REVIEW  
Chemical Formulas and  
Chemical Compounds

SECTION 1 SHORT  
ANSWER Answer the  
following questions in  
the space provided. 1.  
c In a Stock system  
name such as iron(III)  
sulfate, the Roman  
numeral tells us (a)  
how many atoms of Fe  
are in one formula unit.  
(b) how many sulfate  
ions can be attached to  
the iron atom.

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Chemical

**Modern Chemistry  
Chapter 7 Review  
Chemical Formulas  
And ...**

A compound is found to have a molecular mass of 90 atomic mass units and simplest formula of  $C_2H_5O$ .

Question 3 A substance of phosphorus (P) and oxygen (O) is found to have a mole ratio of 0.4 moles of P for every mole of O. The simplest formula for this

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substance is: A. PO<sub>2</sub>B.  
P<sub>0</sub>.4OC. P5O2D. P2O5

## Chemical Formulas Practice Test Questions

MOLECULAR

FORMULAS. The chemical formulas for covalent compounds are referred to as molecular formulas because these compounds exist as separate, discrete molecules. Typically, a molecular formula

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Compounds

begins with the

nonmetal that is  
closest to the lower left  
corner of the periodic  
table, except that  
hydrogen is almost  
never written first ( $H_2O$   
is the prominent  
exception).

## **5.6: Covalent Compounds - Formulas and Names - Chemistry ...**

Start studying  
Chemical Names and  
Formulas Test Review.

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## Chemical

Learn vocabulary, terms, and more with flashcards, games, and other study tools.

Search. ... A reaction where a compound breaks down into elements and or smaller compounds. ... animal farm studyguide answers chapters 1-4. 27 terms. jkhampt4u. Unit 4: Chemical Bonding Test Review. 38 terms.

## **Chemical Names and**

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**Formulas Test  
Review - Quizlet**

Question: Does it matter in what order you combine your elements when writing chemical formulas for ionic compounds? Explain. Ionic Bonds: Ionic compounds are formed by the strong electrostatic ...

**Solved: Does it matter in what order you combine your ...**

Review the chemical

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Compounds

formulas for each compound listed in Data Table 5. Use the Periodic Table of Elements to determine if each element in the compound is a metal or nonmetal and record in Data Table 5. Use the information recorded in Data Tables 1 - 3 to summarize the properties of each substance and record in Data Table 5. Include the state of matter at room



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Formulas  
Compounds

temperature, whether the melting and boiling points were high or low, and whether the compound was soluble or insoluble in water.

## **Solved: Review The Chemical Formulas For Each Compound Lis ...**

SHORT ANSWER 1. a. +4 b. +6 c. -2 d. +1 e. +6 f. +5 g. +4 h. -3 2. a.  $\text{SnCl}_2$  b. nitrogen(IV) oxide 3. a. fluorine b. 0;  $\text{F}_2$  4. a. tin(IV) oxide

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Chemical Formulas and Chemical Compounds

b. SnO 5. a. NO, NO<sub>2</sub> b.  
ClO 6. a. N<sub>2</sub>O<sub>3</sub> b. +3 c.  
The three oxygen  
atoms have oxidation  
states of -6 total, and  
because the algebraic  
sum of the oxidation  
states in a neutral  
compound must be  
zero, the two nitrogen

## **CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds**

Chapter 7 - Chemical  
Formulas & Chemical

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**Chemical Formulas Compounds**  
Compounds Chapter 7 is the final chapter of the first semester. It serves as a summary of all course content discussed so far, as the focus of this chapter is on process skills rather than new concepts or theory.

## **Chapter 7 - Chemical Formulas & Chemical**

### **Compounds - yazvac**

Near the end of class, I lead a brief class

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## Chemical

discussion about ionic and covalent bonding and ask the students to demonstrate their understanding of how to name compounds and write chemical formulas by writing their answers on the board and explaining how they arrived at those answers.

## **And the Name Is....Naming Chemical Compounds**

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Naming Chemical  
Compounds Worksheet  
- Naming Chemical  
Compounds ...

#143705. Naming  
Compounds Handout  
Key - PDF #143706.

The Name Says It All!  
(Naming Compounds  
and Writing Formulas  
... #143707. CHM

130LL:Formula Writing  
and Naming Key

#143708. ... Naming  
Compounds Worksheet  
Answer Key Also

Naming Ionic Pounds ...

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#143720. Naming ions  
and ionic...

## Compounds **Naming compounds worksheet key**

Question: [Review  
Topics] [References]  
Unshared, Or Lone,  
Electron Pairs Play An  
Important Role In  
Determining The  
Chemical And Physical  
Properties Of Organic  
Compounds. Thus, It Is  
Important To Know  
Which Atoms Carry  
Unshared Pairs. Use

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The Structural  
Formulas Below To  
Determine The Number  
Of Unshared Pairs At  
Each Designated Atom.

## **Solved: [Review Topics] [References] Unshared, Or Lone, Ei ...**

Honors Chemistry  
Name Chemical  
Formula Writing  
Review Worksheet #7  
Period Write chemical  
formulas for the  
compounds in each

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Chemical  
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box. The names are found by finding the intersection between the cations and anions.

Example: The first box is the intersection

between the "zinc" cation and the

"chloride" anion, so you should write

"ZnCl<sub>2</sub>", as shown.

chloride acetate nitrate

oxide nitride sulfate

zinc ZnCl<sub>2</sub> iron (II) iron

(III) gallium 10) 11) 12)

gallium GaCl<sub>3</sub>

Ga(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>2</sub> Ga<sub>2</sub>O<sub>3</sub> GaN



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10) 11) 12) silver ...

## Formulas

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e) Ask questions about chemical names to identify patterns in IUPAC nomenclature in order to predict chemical names for ionic (binary and ternary), acidic, and inorganic covalent compounds. f) Develop and use bonding models to predict chemical formulas

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Chemical  
Formulas  
Compounds

including ionic (binary and ternary), acidic, and inorganic covalent compounds.

## **Unit 3 - MHS Honors Chemistry**

Formulas \*\* start studying chemical names and formulas test review learn vocabulary terms and more with flashcards games and other study tools chapter 7 review chemical formulas and chemical compounds

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section 1 short answer  
answer the following  
questions in the space  
provided 1 c in a stock

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